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## Title

Mixtures for producing transparent plastics, transparent plastics and method for their production and use

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The present invention relates to mixtures for producing transparent plastics. Further, the present invention relates to transparent plastics which can be produced from the mixtures, and methods for their production. Further, the present invention relates to the use of transparent plastics for the production of optical, in particular ophthalmic, lenses.

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Nowadays, everyday life without spectacles is no longer imaginable. In particular, spectacles with plastic lenses have recently gained in importance, since they are lighter and less breakable than spectacle lenses of inorganic materials and can be colored with suitable dyes. For the production of plastic spectacle lenses, highly transparent plastics are generally used, which can for example be obtained starting from diethylene glycol bis(allyl)-carbonate (DAC), thiourethane compounds with  $\alpha, \omega$ -terminated multiple bonds or sulfurcontaining (meth)acrylates.

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DAC has very good impact resistance, transparency and good processability. However, a disadvantage is that on account of the relatively low refractive index  $n_D$  of ca. 1.50 both the center and also the edges of the plastic lenses in question have to be reinforced, so that the spectacle lenses are correspondingly thick and heavy. The wearing comfort of spectacles with DAC plastic lenses is therefore markedly diminished.

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The publication DE 4234251 discloses sulfur-containing polymethacrylates, which are obtained by radical copolymer-ization of a monomer mixture of compounds of the formula (1) and (2).

$$\begin{array}{c|c}
R & S \\
\hline
R & S \\
\hline
S & S \\
\hline
\end{array}$$

$$\begin{array}{c|c}
R & S \\
\end{array}$$

$$\begin{array}{c|c}
R$$

Here Y designates for an optionally branched, optionally cyclic alkyl residue with 2 to 12 carbon atoms or an aryl residue with 6 to 14 carbon atoms or an alkaryl residue with 7 to 20 carbon atoms, wherein the carbon chains can be interrupted by one or several ether or thioether groups, R stands for hydrogen or methyl and n is a whole number in the range from 1 to 6.

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According to DE 4234251, the monomers of the formula (1) and (2) are generally in a molar ratio of 1:0.5 to The production of the monomer mixture 0.5:1. least two moles of effected by reaction of at (meth)acryloyl chloride or (meth)acrylic anhydride with one mole of a dithiol, wherein the (meth)acryloyl chloride or (meth)acrylic anhydride are reacted in an inert organic solvent and the dithiol in aqueous solvents, methyl tert.butyl As suitable solution. ether, toluene and xylene are named, the dielectric constants whereof at 20°C are 2.6, 2.4 and 2.3 to 2.6 respectively.

The plastics described in DE 4234251 are colorless, hard and slightly brittle and have a high refractive index  $n_D$  in the range from 1.602 to 1.608. The Abbé number is between 35 and 38. Hence these plastics are also only suitable for spectacle lenses to a limited extent. Information on the glass transition temperature of the plastics can also not be gained from this publication.

In the publication WO 03/011925, the polymerization of thiomethacrylates with polyethylene glycol derivatives is described. The plastics produced from this can inter alia be used for the production of optical lenses. The disadvantage with these lenses is their mechanical properties. Thus in particular the impact resistance does not suffice for many requirements.

In view of the state of the technology, the purpose of to make available the present invention was now 10 mixtures for the production of transparent plastics which are suitable as material for optical lenses, where the plastics have mechanical properties which are good as possible, in particular a high resistance and at the same time a high refractive 15 index, preferably greater than 1.59 and as high as possible an Abbé number, preferably greater than 36. In particular, it should be possible to produce plastic spectacle lenses which display low dispersion and no 20 color fringes.

The present invention was also based on the purpose of composition an educt accessible transparent plastic with production highly of а improved mechanical properties even at temperatures above room temperature. In particular, the plastic according to the invention should have as high as possible a glass transition temperature, preferably greater than 80.0°C.

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A purpose of the present invention was thus to describe a highly transparent plastic, which can be simply and inexpensively produced on the industrial scale starting from the educt composition. In particular, it should be obtainable via free radical polymerization starting from a mixture which is free-flowing at normal pressure and temperatures in the range from 20.0°C to 80.0°C.

The present invention was also based on the purpose of describing application fields and possible uses of the highly transparent plastic according to the invention.

These and other not explicitly mentioned purposes, but which are readily deducible or inferable from the contexts discussed herein, were achieved by means of a mixture with all the features of patent claim 1. Advantageous modifications of the mixture according to invention are placed under protection subclaims referring back to claim 1. In addition to these, the highly transparent plastics obtainable from the mixtures according to the invention and processes for their production are also claimed. The use category a preferred use of the protects the invention. transparent plastic according to optical, preferably ophthalmic lens which contains the highly transparent according to the invention described in a further patent claim.

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## Mixtures comprising

a) a prepolymer produced from compounds of the formula(I) and (II)

wherein R<sup>1</sup> each independently of one another means hydrogen or a methyl residue,

 $R^2$  each independently of one another means a linear or branched, aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue and m and n each independently of one another mean a whole number

greater than or equal to 0 with m + n > 0, and alkyldithiols or polythiols, preferably compounds of the formula (III)  $HS-R^3-SH$  (III)

- wherein  $R^3$  can be the same or different from the meaning given in  $R^2$ 
  - b) at least one radical polymerizable monomer (A) with at least two methacrylate groups and

c) aromatic vinyl compounds,

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which are suitable for the production of transparent plastics and which have outstanding mechanical and optical properties. The mixtures can optionally contain

- d) a radical polymerizable monomer with at least two terminal olefinic groups, which differ in their reactivity, such as is for example the case with a bifunctional monomer with methacrylate terminal group and a vinyl terminal group and/or
- e) at least one ethylenically unsaturated monomer (B), preferably from the group of the methacrylates, especially preferably 2-hydroxyethyl methacrylate.

The transparent plastic according to the invention displays a previously unknown combination of outstanding properties, such as a high refractive index, a high Abbé number, good impact resistance and a high glass transition temperature. The corresponding plastic spectacle lenses show low dispersion; no color fringes can be observed.

- 35 At the same time, the transparent plastic according to the invention has further advantages, such as that:
  - $\Rightarrow$  Because of the high refractive index of the plastic

according to the invention, reinforcement and thus of center the edges of thickening the and spectacle corresponding plastic lenses is not and the wearing comfort of such necessary, spectacles is markedly increased because of the comparatively low weight.

impact resistance of the plastic ⇒ The very good protects according to the invention the plastic spectacle lenses from corresponding "everyday risks". Damage or irreparable destruction, in particular of the thin spectacle lenses through the action of mechanical force is for the most part prevented.

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⇒ The highly transparent plastic according to invention displays high qlass transition a temperature, preferably greater than 80.0°C, and thus retains its outstanding mechanical properties, resistance 20 particular the high impact this temperature. hardness, up to

- highly transparent plastic according to invention can be simply and inexpensively produced industrial scale bv free radical the copolymerization of a monomer mixture which is at normal pressure preferably free-flowing temperatures in the range from 20.0°C to 80.0°C.
- 30 ⇒ The production of the underlying monomer mixture is likewise simply and inexpensively possible on the industrial scale.

The prepolymer of the present invention includes compounds of the formula (I) and/or (II) and (III).

$$\begin{array}{c|c}
R^{I} & & \\
\hline
O & R^{2} & \\
\end{array}$$
(1)

$$\begin{array}{c|c}
R^{1} & R^{2} & R^{2$$

# HS-R3-SH (III),

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wherein the residue R<sup>1</sup> each independently of one another designates hydrogen or a methyl residue, preferably a methyl residue and

independently of one the residue R<sup>2</sup> each designates a linear or branched, aliphatic substituted or residue or а cycloaliphatic hetero-aromatic unsubstituted aromatic or where the residue R<sup>2</sup> can preferably include 1 to 100, in particular 1 to 20 carbon atoms, and

wherein the residue  $R^3$  each independently of  $R^2$  means a linear or branched, aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue, wherein the residue  $R^3$  can preferably include 1 to 100, in particular 1 to 20 carbon atoms.

The preferred linear or branched, aliphatic or cycloaliphatic residues for example include the methylene, ethylene, propylene, iso-propylene, n-butylene, isobutylene, tert.-butylene or cyclohexylene group.

The preferred divalent aromatic or heteroaromatic residues in particular include groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenyl-methane, diphenyldimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention, cyclo-aliphatic residues also include bi-,

tri- and polycyclic aliphatic residues.

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Further, the residue  $R^2$  or  $R^3$  also includes residues of the formula

$$\frac{\left\{\left(R^{4}-X-\right)R^{5}\right\}}{\left(la\right)}$$

wherein  $R^4$  each independently of one another is a linear or branched, aliphatic or cycloaliphatic residue, such as for example a methylene, ethylene, propylene, isopropylene,

iso-butylene, tert.-butylene n-butylene, or 10 The residue X is cyclohexylene group. each independently of one another oxygen or sulfur and the residue R<sup>5</sup> stands for a linear or branched, aliphatic or residue, for example cycloaliphatic such as 15 methylene, ethylene, propylene, iso-propylene, iso-butylene, tert.-butylene n-butylene, group. Also, in the context of cyclohexylene present invention, cycloaliphatic residues also include bi-, tri- and poly-cyclic aliphatic residues. y is a whole number between 1 and 10, in particular 1, 2, 3 20 and 4.

Preferred residues of the formula (Ia) include:

The residue  $R^2$  is preferably an aliphatic residue with 1 to 10 carbon atoms, preferably a linear aliphatic residue with 2 to 8 carbon atoms.

The indices m and n are each independently of one another a whole number greater than or equal to 0, for example 0, 1, 2, 3, 4, 5 or 6. Here the sum m + n is

greater than 0, preferably in the range from 1 to 6, advantageously in the range from 1 to 4, in particular 1, 2 or 3.

The compounds of the formula (I) and (II) and the compounds of the formula (III) respectively can be used individually or as a mixture of several compounds of the formula (I), (II) or (III) for the production of the prepolymer.

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The relative contents of the compounds of the formula (I), (II) and (III) in the monomer mixture according to the invention are in principle arbitrary, and they can be used to "tailor-make" the properties profile of the plastic according to the invention in accordance with the requirements of the use. For example, it can be extremely useful that the monomer mixture contain a marked excess of compound(s) of the formula (I) or compound(s) of the formula (III), each based on the total quantity of the compounds of the formula (I), (II) and (III) in the prepolymer.

However, for the purposes of the present invention it is particularly favorable that the mixture contain more than 10 mol.%, preferably more than 12 mol.%, in particular more than 14 mol.%, based on the total quantity of the compounds of the formula (I) and (II), of compounds of the formula (II) with m + n = 2. If  $R^2$  is an ethylene residue, then the content by weight of (II) with m + n = 2 in the mixture is more than 10%, in particular more than 15%.

Further, it is particularly favorable according to the invention to use mixtures that contain more than 5.8 mol.%, advantageously more than 6.5 mol.%, in particular more than 7.5 mol.%, based on the total quantity of the compounds of the formula (I) and (II),

of compounds of the formula (II) with m + n = 3. This corresponds to a content by weight of (II) with m + n = 3, if  $R^2$  is an ethylene residue, of at least 6%.

The content of the compounds (I) is preferably 0.1 to 5 50.0 mol.%, advantageously 10.0 to 45.0 particular 20.0 to 35.0 mol.%, based on the total quantity of the compounds of the formula (I) and (II), which corresponds to a preferred range of the content by weight of the compound (I), when  $R^2$  is an ethylene 10 residue, of 15 to 40%. The content of the compounds (II) with m + n = 1 is preferably 1 to 40.0 mol.%, advantageously 5 to 35.0 mol.%, in particular 10 to 30 mol.%, based on the total quantity of the compounds of the formula (I) and (II). This corresponds to a content 15 by weight of the compounds (II) with m + n = 1, when  $R^2$ is an ethylene residue, of preferably 10 to 45%. The content of the compounds (II) with m + n > 3preferably greater than 0 mol.%, advantageously greater than 1 mol.%, in particular greater than 2 mol.%, based 20 on the total quantity of the compounds of the formula (I) and (II). If R<sup>2</sup> is an ethylene residue, the content by weight for compounds (II) with m + n > 3 in the mixture is more than 2%, in particular more than 5%.

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The content of the compounds (III) in the prepolymer is preferably 1 to 55.0 mol.%, in particular 10.0 to 50.0 mol.%, based on the total quantity of the compounds of the formula (I), (II) and (III). If in the special case  $R^3$  is a dimercaptodioxaoctane residue, the content by weight of (III) in the prepolymer based on the total quantity of the compounds (I), (II) and (III) is more than 0.5%, preferably more than 5%.

Processes for the production of the compounds of the formulae (I) and (II) are known to the skilled person for example from DE 4234251, reference to the disclosure whereof is hereby explicitly made.

Nonetheless in the context of the present invention it has been found quite especially favorable to produce a mixture of the compounds of the formula (I) and (II) by a process wherein 1.0 to < 2.0 mol, preferably 1.1 to 1.8 mol, advantageously 1.2 to 1.6 mol, in particular 1.2 to 1.5 mol, of at least one compound of the formula (IV)

is reacted with one mole of at least one polyol of the formula (V)

$$M = R^2 = M$$
 (V)

The residue X stands for halogen, in particular chlorine or bromine or for a

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residue, i.e. the compounds of the formula (IV) include inter alia acryloyl chloride, methacryloyl chloride, acrylic anhydride and methacrylic anhydride, the use of acrylic anhydride, methacrylic anhydride or mixtures of the two being particularly preferable.

M each independently of one another designates hydrogen or a metal cation. Preferred metal cations are derived from elements with an electronegativity less than 2.0, advantageously less than 1.5, alkali metal cations, in particular Na $^+$ , K $^+$ , Rb $^+$  and Cs $^+$  and alkaline earth metal cations, in particular Mg $^{2+}$ , Ca $^{2+}$ , Sr $^{2+}$  and Ba $^{2+}$  being particularly preferable. Quite especially favorable results can be achieved with the metal cations Na $^+$  and K $^+$ .

In this connection, particularly suitable polythiols of the formula (V) include 1,2-ethanedithiol, 1,2propanedithiol, 1,3-propanedithiol, 1,2-butanedithiol, 1,3-butanedithiol, 1,4-butanedithiol, 2-methylpropane-1,2-dithiol, 2-methyl-propane-1,3-dithiol, 3,6-dioxa-1,8-octanedithiol (di-mercaptodioxaoctane = DMDO), ethylcyclohexyldimercaptans which are obtainable by reaction of 4-ethenylcyclohexene with hydrogen sulfide, ortho-bis(mercaptomethyl)benzene, metabis(mercaptomethyl)benzene, para-bis(mercaptomethyl)-benzene, compounds of the formula (V)

and compounds of the formula

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$$HS[(R^4-X)R^5]SH$$
 (Va)

where R<sup>4</sup> each independently of one another is a linear or branched, aliphatic or cycloaliphatic residue, such as for example a methylene, ethylene, propylene, iso-propylene, n-butylene, iso-butylene, tert.-butylene or cyclohexylene group. Also, in the sense of the present invention, cyclo-aliphatic residues also include bi-, tri- and polycyclic aliphatic residues. The residue X each independently of one another is oxygen or sulfur

and the residue R<sup>5</sup> stands for a linear or branched, aliphatic or cycloaliphatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or cyclohexylene group. Also, in the sense of the present invention, cycloaliphatic residues also include bi-, tri- and polycyclic aliphatic residues. y is a whole number between 1 and 10, in particular 1, 2, 3 and 4.

10 Preferred compounds of the formula (Va) include:

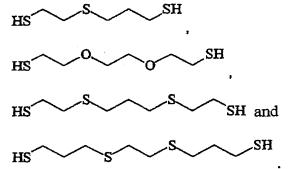
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In the context of a quite especially preferred embodiment of this process, 1,2-ethanedithiol is used as the compound of the formula (V).

According to this process, the compound(s) of the formula (IV) in at least one inert, organic solvent L and the compound(s) of the formula (V) in aqueous-alkaline solution are reacted, the term "inert, organic solvent" standing for such organic solvents as do not react with the compounds present in the reaction system under the given reaction conditions.

Preferably at least one solvent L has a relative dielectric constant > 2.6, preferably > 3.0, advantageously > 4.0, in particular > 5.0, each measured at 20°C. In this connection, the relative dielectric constant designates a dimensionless number which states by how many times the capacity C of a condenser (theoretically) situated in a vacuum is increased when substances with dielectric properties, so-called dielectrics, are placed between the plates.

This value is measured at 20°C and extrapolated to low frequencies  $(\omega \rightarrow 0)$ . For further details, reference is specialist literature, current made to the Industrial to Ullmann's Encyclopedia of particular Application of physical 2/1 5 Chemistry, Vol. physical chemical methods in the laboratory, keyword: Dielectric constants, pp. 455-479. Dielectric constants of solvents are inter alia quoted in the Handbook of Chemistry and Physics, 71st Edition, CRC Press, Baco Raton, Ann Arbor, Boston, 1990-1991, pp. 8-44, 8-46 and 10 9-9 to 9-12.

In the context of this process, it is further particularly advantageous if the solvent and the aqueous solution form two phases during the reaction and are not homogeneously miscible. For this purpose, the solvent preferably has a water solubility, measured at 20°C, less than 10 g of water based on 100 g of solvent.

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Solvents L preferred according to the invention include aliphatic ethers such as diethyl ether (4.335), dipropyl ether, diisopropyl ether, cycloaliphatic ethers such as tetrahydrofuran (7.6);

aliphatic esters such as methyl formate (8.5), ethyl 25 formate, propyl formate, methyl acetate, ethyl acetate, (5.01), methyl propionate, methyl n-butyl acetate butyrate (5.6), ethyl butyrate, 2-methoxyethyl acetate; aromatic esters such as benzyl acetate, dimethyl 30 phthalate, methyl benzoate (6.59),ethyl benzoate ethyl salicylate, (6.02), methyl salicylate, phenyl acetate (5.23);

aliphatic ketones such as acetone, methyl ethyl ketone (18.5), pentan-2-one (15.4), pentan-3-one (17.0),

methyl isoamyl ketone, methyl isobutyl ketone (13.1); aromatic ketones such as acetophenone; nitroaromatics such as nitrobenzene, o-nitrotoluene (27.4), m-nitrotoluene (23), p-nitrotoluene;

halogenated aromatics such as chlorobenzene (5.708), o-chlorotoluene (4.45), m-chlorotoluene (5.55), p-chloro-toluene (6.08), o-dichlorobenzene, m-dichlorobenzene;

heteroaromatics such as pyridine, 2-methylpyridine (9.8), quinoline, isoquinoline; or mixtures of these compounds, where the information the brackets are the respective applicable relative dielectric constants at 20°C.

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Here, for the purposes of the present process, aliphatic esters and cycloaliphatic esters, in particular ethyl acetate and tetrahydrofuran are quite especially suitable.

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In the context of the present invention, the solvent L can be used both alone and also as a solvent mixture, wherein not all solvents contained in the mixture have fulfill the above dielectricity criterion. example, tetrahydro-furan/cyclohexane mixtures can also 20 be used according to the invention. However, that the solvent mixture found advantageous dielectric constant > relative а displays preferably > 3.0, advantageously > 4.0, in particular > each measured at 20°C. Especially advantageous 25 results can be achieved with solvent mixtures which only contain solvents with a relative dielectric constant > 2.6, preferably > 3.0, advantageously > 4.0, in particular > 5.0, each measured at 20°C.

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The aqueous alkaline solution of the compound(s) of the formula (V) preferably contains 1.1 to 1.5 val (equivalents) of at least one Bronsted base, based on the total quantity of compound(s) of the formula (IV). Preferred Bronsted bases in the sense of the present invention include alkali metal hydroxides and alkaline earth metal hydroxides, in particular sodium hydroxide and potassium hydroxide.

The reaction can in principle be performed in any possible manner, for example it is possible to take the of the formula (IV) in the compound(s) (mixture) L and add the aqueous-alkaline solution of compound(s) of the formula (V) stepwise continuously. However, in the context of the present invention it has been found particularly favorable to feed the compound(s) of the formula (IV) in at least one inert, organic solvent L and the compound(s) of the in aqueous-alkaline solution formula (V) reaction vessel in parallel.

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The reaction temperature can be varied over a wide range, but the temperature often lies in the range from 20.0°C to 120.0°C, preferably in the range from 20.0°C to 80.0°C. The same applies for the pressure at which the reaction is carried out. Thus the reaction can take place both under decreased pressure and also under increased pressure. It is however preferably performed at normal pressure. Although the reaction can also take place in air, it has been found particularly favorable in the context of the present invention to perform the reaction under a protective gas atmosphere, preferably nitrogen and/or argon, wherein a small oxygen content is preferably present.

Advantageously, the reaction mixture is treated with a Bronsted acid in a further step, preferably until the aqueous solution at 20°C displays a pH value less than 7.0, advantageously less than 6.0, in particular less in this connection include 5.0. Usable acids inorganic mineral acids such as hydrochloric acid, sulfuric acid and phosphoric acid, organic acids such acid and propionic acid, and acetic acid exchangers, in particular acid synthetic resin exchangers, such as for example \*Dowex M-31 (H). Here the use of acid synthetic resin ion exchangers with loadings of at least 1.0 meq, preferably at least 2.0 meq, in particular at least 4.0 meq, H<sup>+</sup> ions based on 1 g of dried ion exchanger, particle sizes of 10-50 mesh and porosities in the range from 10 to 50% based on the total volume of the ion exchanger has proved quite especially effective.

For the isolation of the compounds of the formula (I) and (II), the organic phase consisting of the solvent L is advantageously separated, washed if necessary, dried and the solvent evaporated.

In the reaction of the compound(s) of the formula (IV) with the compound(s) of the formula (V), inhibitors can be added, which prevent radical polymerization of the (meth)acryl groups during the reaction. These inhibitors are widely known in the specialist field.

1,4-dihydroxybenzene is mainly used. However, differently substituted dihydroxybenzenes can also be used. In general such inhibitors can be described by the general formula (VI)

$$R^7O$$
—OH (VI),

wherein

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25 R<sup>6</sup> is a linear or branched alkyl residue with one to eight carbon atoms, halogen or aryl, preferably an alkyl residue with one to four carbon atoms, particularly preferably methyl, ethyl, n-propyl, isopropyl, n-butyl, iso-butyl, sec-butyl, tert.-butyl, Cl, 30 F or Br;

o is a whole number in the range from one to four, preferably one or two; and

R<sup>7</sup> means hydrogen, a linear or branched alkyl residue 35 with one to eight carbon atoms or aryl, preferably an alkyl residue with one to four carbon atoms, particularly preferably methyl, ethyl, n-propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl or tert.-butyl.

However compounds with 1,4-benzoquinone as the parent compound can also be used. These can be described by the formula (VII)

$$0 = \bigcap_{R_0^6} 0$$
 (VII)

wherein

 $R^6$  and o have the meaning stated above.

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Likewise phenols of the general structure (VIII) are used

wherein

15 R<sup>8</sup> means a linear or branched alkyl residue with one to eight carbon atoms, aryl or aralkyl, propionate ester with 1 to 4-hydric alcohols, which can also contain hetero atoms such as S, O and N, preferably an alkyl residue with one to four carbon atoms, particularly preferably methyl, ethyl, n-propyl, iso-propyl, n-butyl, iso-butyl, sec-butyl or tert.-butyl.

A further advantageous class of substances is represented by hindered phenols based on triazine derivatives of the formula (IX)

with  $R^9$  = compound of the formula (X)

wherein  $R^{10} = C_p H_{2p+1}$  with p = 1 or 2.

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The compounds 1,4-dihydroxybenzene, 4-methoxyphenol, 5 2,5-dichloro-3,6-dihydroxy-1,4-benzoquinone, 1,3,5trimethyl-2,4,6-tris-(3,5-di-tert.-butyl-4-2,6-di-tert.-butyl-4hydroxybenzyl) benzene, 2,4-dimethyl-6-tert.-butylphenol, 2,2methylphenol, bis[3,5-bis(1,1-dimethylethyl)-4-hydroxy-phenyl-1-10 2,2'-thioester, oxopropoxymethyl)]1,3-propandiyl diethylbis-[3-(3,5-di-tert.butyl-4-hydroxyphenyl)] octadecyl-3-(3,5-di-tert.-butyl-4prop-ionate, hydroxyphenyl) propionate, 3,5-bis(1,1-dimethylethyl-2,2-methylenebis-(4-methyl-6-tert.-butyl)phenol, 15 (4-tert.-butyl-3-hydroxy-2,6-dimethylbenzyl)-s-triazintris-(3,5-di-tert.butyl-4-2,4,6-(1H,3H,5H) trione, hydroxy)-s-triazin-2,46-(1H,3H,5H)-trione or tert.butyl-3,5-dihydroxybenzene particularly are 20 successfully used.

Based on the weight of the whole reaction mixture, the content of the inhibitors alone or as a mixture is in general 0.01 to 0.50 % (wt/wt), the concentration of the inhibitors preferably being selected such that the color number according to DIN 55945 is not affected. Many of these inhibitors are commercially available.

In the context of the present invention, in addition to the prepolymer produced from compounds of the formula (I), (II) and (III), the mixture further contains at least one radical polymerizable monomer (A) with at least two terminal methacrylate groups.

35 Di(meth)acrylates coming under this heading are for example

polyoxyethylene and polyoxypropylene derivatives such as triethylene (meth)acrylic acid, (meth) acrylate, tetraethylene glycol (meth)acrylate, (meth) acrylate, glycol and tetra-propylene diethylene di (meth) acrylate, qlycol 5 butanediol di(meth)acrylate, dipropylene glycol di(meth)acrylate, di(meth)acrylate, glycol tripropylene triethylene di(meth)acrylate, glycol tetraethylene glycol di (meth) acrylate, tetrapropylene glycol 10 di (meth) acrylate, polyethylene glycol di (meth) -acrylate (preferably with weight averaged molecular weights in the range from 200 to 5,000,000 g/mol, advantageously in the range from 200 to 25,000 g/mol, in particular in the range from 200 to 1000 g/mol), polypropylene glycol di(meth)acrylate (preferably with weight averaged 15 molecular weight in the range from 200 to 5,000,000 g/mol, advantageously in the range from 250 to 4000 g/mol, in particular in the range from 250 to 1000 g/mol), 2,2'-thioethanol di(meth)acrylate (thiodiglycol 20 (meth) acrylate),

3,9di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,
in particular

3,8-

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di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,
4,8-

di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,
4.9-

di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane, ethoxylated bisphenol A di(meth)acrylate, in particular

wherein s and t are greater than or equal to zero and

the sum s + t preferably lies in the range from 1 to 30, in particular in the range from 2 to 10, and di(meth)acrylates obtainable by reaction of diisocyanates with 2 equivalents of hydroxyalkyl (meth)acrylate, in particular

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wherein the residue R<sup>11</sup> each independently of one another means hydrogen or a methyl residue, tri(meth)acrylates, such as trimethylolpropane tri(meth)-acrylate and glycerin tri(meth)acrylate or else (meth)acrylates of ethoxylated or propoxylated glycerin, trimethylolpropane or other alcohols with more than two hydroxy groups.

As monomers (A), di(meth)acrylates of the formula (XI)

$$\begin{array}{c}
\mathbb{R}^{12} \\
\mathbb{R}^{13} \\
\mathbb{R}^{13}
\end{array}$$
(XI),

have proved particularly effective. Here  $R^{12}$  each independently of one another means hydrogen or methyl.  $R^{13}$  designates a linear or branched alkyl or cycloalkyl residue or an aromatic residue with preferably 1 to 100, preferably 1 to 40, preferably 1 to 20, advantageously 1 to 8, in particular 1 to 6 carbon atoms, such as for example a methyl, ethyl, propyl,

iso-butyl, tert-butyl, iso-propyl, n-butyl, cyclopentyl, cyclohexyl or phenyl group. Also, in the invention present cycloaliphatic of the and polycyclic include bi-, triresidues also Linear or branched alkyl aliphatic residues. cycloalkyl residues with 1 to 6 carbon atoms are quite particularly preferred as R18.

The residue R<sup>13</sup> is preferably a linear or branched, aliphatic or cycloaliphatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or cyclohexylene group or a residue of the general formula (XIa)

 $\frac{[R^{14} X^{I}]R^{15}}{[R^{14} X^{I}]R^{15}}$ 

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wherein the residue  $R^{15}$  stands for a linear or branched, aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue, as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or divalent aromatic or cvclohexylene group, or heteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyl-dimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic include bi-, triand polycyclic also residues aliphatic residues. The residue R14 each independently a linear or another designates aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or cyclohexylene or divalent aromatic group, heteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyldimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic residues also include bi-, tri- and polycyclic aliphatic residues. The residue  $X^1$  is each independently of one another oxygen, sulfur, an ester group of the general formulae (XIb) or (XIc)

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an urethane group of the general formulae (XId), (XIe), (XIf) or (XIg),

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a thiourethane group of the general formula (XIh), (XIi), (XIj) or (XIk),

a dithiourethane group of the general formula (XII), (XIm), (XIn) or (XIo),

or a thiocarbamate group of the general formula (XIp), (XIq), (XIr) or (XIs),

R<sup>16</sup> S (XIs)

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preferably oxygen, wherein the residue  $R^{16}$  stands for a linear or branched, aliphatic or cycloaliphatic residue substituted or unsubstituted aromatic heteroaromatic residue, such as for example a methyl, ethyl, propyl, iso-propyl, n-butyl, iso-butyl, tert.butyl or cyclohexyl group, or monovalent aromatic or heteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyldimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic residues also include bi-, triand polycyclic aliphatic residues. z is a whole number between 1 and 1000, advantageously between 1 and 100, in particular between 1 and 25.

Particularly preferred di(meth)acrylates of the formula (XI) include ethylene glycol di(meth)acrylate, ethoxylated bisphenol A di(meth)acrylate, in particular

wherein s and t are greater than zero and the sum s + t preferably lies in the range from 1 to 20, in particular in the range from 2 to 10,

and di(meth)acrylates obtainable by reaction of diisocyanates with two equivalents of hydroxyalkyl (meth)acrylate, in particular

$$\begin{array}{c} \begin{array}{c} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \begin{array}{c} \\ \\ \\ \\ \\ \end{array} \begin{array}{c} \\$$

wherein the residue  $R^{17}$  each independently of one another means hydrogen or a methyl residue,

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di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,
3,9-

di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,
4,8-

di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane,

di(meth)acryloyloxymethyltricyclo[5.2.1.0(2.6)]decane, thioglycol di(meth)acrylate, polypropylene glycol di(meth)-acrylate, preferably with a weight averaged molecular weight in the range from 200 to 1000 g/mol

and/or poly-ethylene glycol di (meth) acrylate, preferably with a weight averaged molecular weight in 200 to the range from dimethacrylates of the Here the said 1000 q/mol. particularly preferable. compounds are particularly advantageous results are achieved with the use of polyethylene glycol di(meth)acrylate, preferably with a weight averaged molecular weight in the range from 200 to 1000 g/mol.

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The content of monomer (A) is 2-50 wt.%, in particular 10-30 wt.%, based on all monomers used in the mixture.

In the context of the present invention, in addition to the prepolymer, consisting of the compounds of the formula (I), (II) and (III) and at least one radical polymerizable monomer (A), the mixture also contains an aromatic vinyl compound.

20 aromatic vinyl compounds, styrenes, Among the substituted styrenes with an alkyl substituent in the side-chain, such as for example  $\alpha$ -methylstyrene and  $\alpha$ with ethylstyrene, substituted styrenes an substituent on the ring, such as vinyltoluene and p-25 methylstyrene, halogenated styrenes, such monochlorostyrene, dichloro-styrene, example tribromostyrene and tetrabromostyrene

and dienes, such as for example 1,2-divinylbenzene,

1,3-divinylbenzene, 1,4-divinylbenzene, 1,2diisopropenyl-benzene, 1,3-diisopropenylbenzene and
1,4-diisopropenyl-benzene, are preferably used.

The content of the aromatic vinyl compounds is 5 to 40 wt.%, preferably 10 to 30 wt.%, particularly preferably 15 to 25 wt.%, based on the total quantity of the compounds of the formula (I), (II) and (III) which are used in the prepolymer, the radical

polymerizable monomer (A) and the aromatic vinyl compounds and other optionally used monomers.

Surprisingly, the addition of monomer (A) and the aromatic vinyl compound improves the mechanical properties of the plastic material according to the invention without adversely affecting its optical properties. In many cases, a favorable effect on the optical properties can be noted.

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According to a particular aspect of the present invention, compounds preferably linearly structured molecules of different chain length (asymmetric crosslinkers) of the general formula (XII) can be contained

$$Y = R^{18}$$

$$R^{19}$$
(XII),

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wherein the residue R<sup>19</sup> independently means a hydrogen atom, a fluorine atom and/or a methyl group, residue R18 means a linking group, which preferably includes 1 to 1000, in particular 2 to 100 carbon atoms, and the residue Y means a bond or a linking group with 0 to 1000 carbon atoms, in particular 1 to 1000 carbon atoms, and preferably 1 to 100 carbon atoms. The length of the molecule can be varied via the molecule part R18. Compounds of the formula (XII) have a terminal (meth)acrylate function at one end of the group different and a terminal from a molecule, methacrylate function at the other. The preferred groups Y include in particular a bond (vinyl group) a CH<sub>2</sub> group (allyl group) and aliphatic or aromatic groups with 1 to 20 carbon atoms, such as for example a group derived from benzene, wherein particularly preferably the aliphatic or aromatic groups contain a urethane group.

35 The residue R<sup>18</sup> is preferably a linear or branched, aliphatic or cycloaliphatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or cyclohexylene group or a residue of the general formula  $\frac{\{(X^{20}X^1)_{i}R^{21}\}}{\{(X^{10}X^1)_{i}R^{21}\}}$ 

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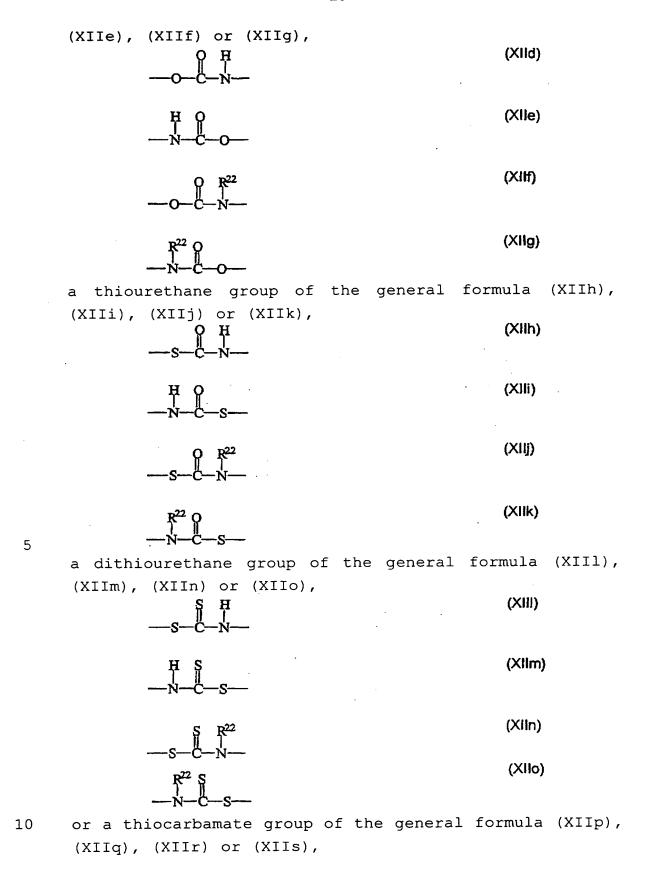
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wherein the residue R<sup>21</sup> stands for a linear or branched, aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or aromatic cvclohexylene group, or divalent orheteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyl-dimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic also include bi-, triand aliphatic residues. The residue R<sup>20</sup> each independently one another designates a linear or branched, aliphatic or cycloaliphatic residue or a substituted or unsubstituted aromatic or heteroaromatic residue, such as for example a methylene, ethylene, propylene, isopropylene, n-butylene, iso-butylene, tert.-butylene or divalent cyclohexylene group, or aromatic heteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyldimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic polycyclic include bi-, triand residues also aliphatic residues. The residue X1 each independently of one another is oxygen, sulfur, an ester group of the general formula (XIIb) or (XIIc)

an urethane group of the general formulae (XIId),



preferably oxygen, wherein the residue R<sup>22</sup> stands for a linear or branched, aliphatic or cycloaliphatic residue substituted or unsubstituted aromatic or heteroaromatic residue, such as for example a methyl, ethyl, propyl, iso-propyl, n-butyl, iso-butyl, tert.butyl or cyclohexyl group, or monovalent aromatic or heteroaromatic groups which are derived from benzene, naphthalene, biphenyl, diphenyl ether, diphenylmethane, diphenyldimethylmethane, bisphenone, diphenyl sulfone, quinoline, pyridine, anthracene and phenanthrene. Also, in the context of the present invention cycloaliphatic residues also include bi-, triand polycyclic aliphatic residues. z is a whole number between 1 and 1000, advantageously between 1 and 100, in particular between 1 and 25.

A particular embodiment of the formula (XII) concerns compounds of the formula (XIII)

$$\begin{array}{c|c}
R^{23} & & \\
\hline
Q & R^{25} & \\
\hline
Q & R^{24} & Q
\end{array}$$
(XIII)

and/or of the formula (XIV)

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$$\begin{array}{c|c}
R^{23} & O & O \\
\hline
 & R^{24} & R^{23}
\end{array}$$
(XIV)

wherein the residues  ${\ensuremath{\text{R}}}^{23}$  and  ${\ensuremath{\text{R}}}^{24}$  each independently of

one another stand for hydrogen or a methyl residue, and the residue  $R^{25}$  designates a linear or branched, aliphatic or cycloaliphatic divalent residue or a substituted or unsubstituted aromatic or heteroaromatic divalent residue. Preferred residues have been presented above.

The length of the chain can be influenced by variation of the number of polyalkylene oxide units, preferably poly-ethylene glycol units. Compounds of the formulae (XIII) and (XIV), which for r, p and q independently of one another have 1-40, preferably 5-20, in particular 7 to 15 and quite particularly preferably 8-12 polyalkylene oxide units, have been found particularly suitable for the solution of the problem described here.

Asymmetric crosslinkers quite particularly preferred according to the invention include compounds of the formula (XIV), in particular

wherein s and t are greater than or equal to zero and the sum s + t preferably lies in the range from 1 to 20, in particular in the range from 2 to 10, and compounds of the formula (XIII), in particular

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(XIVb)

wherein s and t are greater than or equal to zero and the sum s + t preferably lies in the range from 1 to 20, in particular in the range from 2 to 10.

According to a particular aspect, the mixture preferably contains 0.5-40 wt.%, in particular 5 to 15

wt.% of compounds of the formula (XII) and/or (XIII), based on the total weight of the monomer mixture.

In the context of a particularly preferred embodiment of the present invention, the mixture according to the invention additionally contains at least ethylenically unsaturated monomer (B). These monomers differ asymmetric compounds of from the formulae (XIII) and (XIV), the monomers (A) and the thio(meth)acrylates of the formulae (I) and/or The monomers (B) are known in the specialist field and are preferably copolymerizable with the monomers (A) and the thio(meth)acrylates of the formulae (I) and/or (II). These monomers (B) include in particular:

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Nitriles of (meth)acrylic acid and other nitrogencontaining methacrylates, such as methacryloylamidoacetonitrile, 2-methacryloyloxyethylmethylcyanamide and cyanomethyl methacrylate;

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(Meth)acrylates which are derived from saturated alcohols, such as methyl (meth)acrylate, ethyl (meth)acrylate,

n-propyl (meth)acrylate, iso-propyl (meth)acrylate, nbutyl (meth)acrylate, sec-butyl (meth)acrylate, tert.butyl (meth)acrylate, pentyl (meth) acrylate, hexyl (meth)-acrylate, heptyl (meth)acrylate, 2-ethylhexyl (meth) -acrylate, octyl (meth)acrylate, (meth) acrylate, (meth)acrylate, iso-octyl iso-nonyl (meth)acrylate, 2-tert.-butylheptyl (meth)acrylate, 3iso-propylheptyl (meth)-acrylate, decyl (meth)acrylate, undecyl (meth)acrylate, 5-methylundecyl (meth)acrylate, dodecyl (meth)acrylate, 2-methyldodecyl (meth)acrylate, 5-methyltridecyl tridecyl (meth) acrylate, (meth)acrylate, tetradecyl (meth)acrylate, pentadecyl (meth) acrylate, (meth) acrylate, hexadecyl methylhexadecyl (meth)acrylate, heptadecyl (meth)-

acrylate, 5-iso-propylheptadecyl (meth)acrylate,

tert.-butyloctadecyl (meth)acrylate, 5-ethyloctadecyl (meth)-acrylate, 3-isopropyloctadecyl (meth)acrylate, octadecyl (meth)acrylate, nonadecyl (meth)acrylate, eicosyl (meth)-acrylate, cetyleicosyl (meth)acrylate, stearyleicosyl (meth)acrylate, docosyl (meth)acrylate and/or eicosyl-tetratriacontyl (meth)acrylate;

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Cycloalkyl methacrylates, such as cyclopentyl (meth)-acrylate, cyclohexyl (meth)acrylate, 3-vinyl-2-butyl-cyclohexyl (meth)acrylate and bornyl (meth)acrylate;

(Meth)acrylates which are derived from unsaturated alcohols, such as 2-propynyl (meth)acrylate, allyl (meth)-acrylate and oleyl (meth)acrylate and vinyl (meth)acrylate;

Aryl (meth)acrylates, such as benzyl (meth)acrylate or phenyl (meth)acrylate, wherein the aryl residues can in each case be unsubstituted or up to four times substituted;

Hydroxyalkyl (meth)acrylate, such as 3-hydroxypropyl (meth)acrylate, 3,4-dihydroxybutyl (meth)acrylate, 2-hydroxypropyl (meth)-acrylate, 2,5-dimethyl-1,6-hexanediol (meth)acrylate, 1,10-decanediol (meth)acrylate and 1,2-propanediol (meth)-acrylate;

Aminoalkyl (meth)acrylate, such as tris(2-methacryloxy-30 ethyl)amine, N-methylformamidoethyl (meth)acrylate and 2-ureidoethyl (meth)acrylate;

Carbonyl-containing (meth)acrylate, such as 2-carboxyethyl (meth)acrylate, carboxymethyl (meth)acrylate, oxazolidinyl-ethyl (meth)acrylate, N-(methacryloyloxy)formamide, acetonyl (meth)acrylate, N-methacryloylmorpholine and N-methacryloyl-2-pyrrolidinone;

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(Meth) acrylates of ether alcohols such as tetrahydro-
                (meth)acrylate, vinyloxyethoxyethyl
      furfuryl
                                                        (meth) -
      acrylate,
                methoxyethoxyethyl (meth)acrylate, 1-butoxy-
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                 (meth) acrylate,
                                    1-methyl-(2-vinyloxy)ethyl
      propyl
                         cyclohexyloxymethyl
                                                (meth) acrylate,
      (meth) -acrylate,
      methoxy-methoxyethyl
                             (meth) acrylate,
                                                benzyloxymethyl
      (meth) -acrylate, furfuryl (meth) acrylate, 2-butoxyethyl
      (meth) -acrylate,
                         2-ethoxyethoxymethyl
                                                (meth) acrylate,
      2-ethoxy-ethyl
                                                 allyloxymethyl
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                           (meth) acrylate,
      (meth) acrylate,
                                                (meth) acrylate,
                            1-ethoxybutyl
      methoxymethyl
                           (meth) acrylate,
                                                  1-ethoxyethyl
      (meth)acrylate and ethoxymethyl (meth)-acrylate;
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      (Meth)acrylates of halogenated alcohols, such as 2,3-
      dibromopropyl
                           (meth) acrylate,
                                                  4-bromophenyl
                        1,3-dichloro-2-propyl (meth)acrylate,
      (meth) acrylate,
                                                    2-iodoethyl
      2-bromoethyl
                           (meth) -acrylate,
      (meth) acrylate and chloromethyl (meth) acrylate;
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                                                 2,3-epoxybutyl
      Oxiranyl
                 (meth) acrylates,
                                     such
                                            as
                         3,4-epoxybutyl
      (meth) acrylate,
                                           (meth) acrylate
                                                             and
      glycidyl (meth)acrylate;
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      Amides of (meth)acrylic acid, such as
      N-(3-dimethylaminopropyl) (meth)acrylamide,
      N-(diethylphosphono)(meth)acrylamide,
      1-(meth)acryloylamido-2-methyl-2-propanol,
      N-(3-dibutylaminopropyl) (meth) acrylamide,
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      N-t-butyl-N-(diethylphosphono) (meth) acrylamide,
      N, N-bis (2-diethylaminoethyl) (meth) acrylamide,
      4-(methyl)acryloylamido-4-methyl-2-pentanol,
      N-(methoxymethyl)(meth)acrylamide,
      N-(2-hydroxyethyl) (meth) acrylamide,
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      N-(acetyl) (meth) acrylamide,
      N-(dimethylaminoethyl) (meth) acrylamide,
      N-methyl-N-phenyl (meth) acrylamide,
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N, N-diethyl (meth) acrylamide, N-methyl (meth) acrylamide,

N,N-dimethyl(meth)acrylamide and
N-isopropyl(meth)acrylamide;

Heterocyclic (meth)acrylates, such as 2-(1-imidazolyl)
ethyl (meth)acrylate, 2-(4-morpholinyl)ethyl

(meth)acrylate and 1-(2-methacryloyloxyethyl)-2
pyrrolidone;

Phosphorus, boron and/or silicon-containing

(meth)acrylates such as
2-(dimethylphosphonato)propyl (meth)acrylate,
2-(ethylenphosphito)propyl (meth)acrylate,
dimethylphosphinomethyl (meth)acrylate,
dimethylphosphonoethyl (meth)acrylate,
diethyl(meth)acryloyl phosphonate and
dipropyl(methacryloyl) phosphate;

Sulfur-containing (meth) acrylates such as ethylsulfinylethyl (meth) acrylate, 4-thiocyanatobutyl (meth) acrylate, ethylsulfonylethyl (meth) acrylate, thiocyanatomethyl (meth) acrylate, methylsulfinylmethyl (meth) acrylate and bis((meth) acryloyloxyethyl) sulfide;

Bis(allyl carbonates) such as ethylene glycol bis(allyl carbonate), 1,4-butanediol bis(allyl carbonate) and diethylene glycol bis(allyl carbonate);

Vinyl halides, such as for example vinyl chloride, vinyl fluoride, vinylidene chloride and vinylidene fluoride;

Vinyl esters such as vinyl acetate;

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Heterocyclic vinyl compounds such as 2-vinylpyridine,
3-vinylpyridine, 2-methyl-5-vinylpyridine, 3-ethyl-4vinyl-pyridine, 2,3-dimethyl-5-vinylpyridine,
vinylpyrimidine, vinylpiperidine, 9-vinylcarbazole, 3vinylcarbazole,

4-vinylcarbazole, 1-vinylimidazole, 2-methyl-1-vinyl-imidazole, N-vinylpyrrolidone, 2-vinylpyrrolidone, N-vinyl-pyrrolidine, 3-vinylpyrrolidine, N-vinylcaprolactam,

- N-vinylbutyrolactam, vinyloxolan, vinylfuran, vinylthiophen, vinylthiolan, vinylthiazoles and hydrogenated vinylthiazoles, vinyloxazoles and hydrogenated vinyloxazoles;
- 10 Vinyl and isoprenyl ethers;

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Maleic acid and maleic acid derivatives, such as for example mono- and diesters of maleic acid, wherein the alcohol residues have 1 to 9 carbon atoms;

Maleic anhydride, methylmaleic anhydride, maleimide and methylmaleimide;

Fumaric acid and fumaric acid derivatives, such as for example mono- and diesters of fumaric acid, wherein the alcohol residues have 1 to 9 carbon atoms.

In addition, a di(meth)acrylate listed under monomer (A) can also be used as monomer (B).

In this connection, the expression (meth) acrylates includes methacrylates and acrylates and mixtures thereof. Likewise, the expression (meth) acrylic acid includes methacrylic acid and acrylic acid and mixtures thereof.

The ethylenically unsaturated monomers can be used singly or as mixtures.

The composition of the monomer mixtures according to the invention is in principle any desired. It can be used so as to adapt the property profile of the plastic according to the invention to the requirements of the

use. It has however been found extremely advantageous to select the composition of the monomer mixture such that the prepolymer consisting of the compound(s) of (I), (II) and (III)and at least formula and styrene mix homogeneously monomer (A) temperature, since such desired polymerization mixtures, on account inter alia of their low viscosity, are easily manageable and moreover can be polymerized plastics with improved homogeneous properties.

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According to a particularly preferred embodiment of the present invention, the monomer mixture contains a prepolymer consisting of at least 5.0 wt.%, preferably at least 20.0 wt.%, particularly preferably at least 50.0 wt.% of compounds of the formula (I), (II) and (III), each based on the total weight of the monomer mixture. The content by weight of the monomer (A) is preferably at least 2.0 wt.%, preferably at least 10.0 wt.%, particularly preferably at least 20.0 wt.%, each based on the total weight of the monomer mixture. The content by weight of aromatic vinyl compounds, in particular styrene, is preferably at least 2.0 wt.%, preferably at least 10.0 wt.%, particularly preferably at least 20.0 wt.%, preferably at least 20.0 wt.%, preferably at least 20.0 wt.%, particularly preferably at least 20.0 wt.%, each based on the total weight of the monomer mixture.

a particular aspect of the present According to invention, the mixture contains 50 to 90 wt.8, in 60 to particular 85 wt.% of the prepolymer of the monomers of the formulae (I) and/or (II) and (III), 2 to 50 wt.%, in particular 10 to 30 wt.% of monomers (A) and 2 to 50 wt.%, in particular 10 to 30 wt.% of vinyl compounds, in particular styrene and 0 to 45 wt.%, in particular 1 to 10 wt.% of monomers of the formulae (XII) and (XIII) and/or monomers (B), each based on the total weight of the monomer mixture.

The production of the monomer mixture to be used according to the invention is well known to the skilled person. It can for example be effected by mixing of the prepolymer, consisting of thio(meth)acrylates of the formulae (I) and/or (II) with (III), the aromatic vinyl compounds and the monomers (A) and (B) in a manner known per se.

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10 For the purposes of the present invention, the monomer mixture is preferably free-flowing at normal pressure and temperatures in the range from 20.0°C to 80.0°C. The term "free-flowing" is well known to the person. It designates a liquid which preferably can be poured into various molds and stirred and homogenized 15 with the aid of suitable additives. Particular, freeflowing compounds in the sense of the invention have in particular dynamic viscosities of the order of 0.1 mPa.sec to 10 Pa.sec, advantageously in the range from 20 0.65 mPa.sec to 1 Pa.sec at 25°C and at normal pressure Pa). quite particularly In а embodiment of the present invention, a poured monomer mixture contains no bubbles, in particular, no air bubbles. Also preferred are those monomer mixtures out 25 of which bubbles, in particular air bubbles, can be removed by suitable processes, such as for example increasing the temperature and/or application of a vacuum.

highly plastic 30 transparent according the The to invention is obtainable by free copolymerization of the low viscosity ( $\eta < 200 \text{ mPa.sec}$ ) mixture described above. Free monomer copolymerization is a widely known process initiated by 35 free radicals, wherein a mixture of low molecular weight monomers is converted into high molecular weight compounds, so-called polymers. For further details, reference is made to the disclosure by H G Elias,

Macromolecules, Vols. 1 and 2, Basel, Heidelberg, New York Hüthig and Wepf. 1990 and Ullmann's Encyclopedia of Industrial Chemistry, 5<sup>th</sup> Edn., Keyword "Polymerization Processes".

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In a preferred embodiment of the present invention, the plastic according to the invention is obtainable by solventless polymerization of the monomer mixture. Here, bulk or solventless polymerization is understood to mean a polymerization process wherein monomers are polymerized without a solvent, so that the polymerization reaction takes place in bulk in Polymerization emulsion (so-called solventless. polymerization) and polymeriz-ation emulsion polymerization), (so-called suspension dispersion the organic monomers are suspended in wherein and/or colloids protective aqueous phase with less coarse polymerizate stabilizers and more or particles are formed, are to be regarded antithesis of this. A particular form of polymerization in heterogeneous phase is bead polymerization, which can essentially be considered as a type of suspension polymerization.

The polymerization reaction can basically be triggered in any way familiar to the skilled person, for example initiated with the use of a radical initiator (e.g. peroxide, azo compound) or by irradiation with UV radiation, visible light or  $\alpha$  rays,  $\beta$  rays or  $\gamma$  rays, or by a combination thereof.

In a preferred embodiment of the present invention, lipophilic radical polymerization initiators are used for the initiation of the polymerization. The radical polymer-ization initiators are lipophilic in particular so that they dissolve in the solventless polymerization mixture. Usable compounds, apart from the classical azo initiators such as azoisobutyronitrile (AIBN) or 1,1-

azobiscyclohexan-carbonitrile, include inter alia compounds such as for example peroxy tert.-amyl peroxyneodecanoate, tert.-amyl peroxypivalate, tert.peroxypivalate, tert.-amyl peroxv-2ethylhexanoate, tert.-butyl peroxy-2-ethyl-hexanoate, tert.-amyl peroxy-3,5,5-trimethylhexanoate, ethyl 3,3di-(tert.-amylperoxy)-butyrate, tert.-butyl tert.-butyl hydroperoxide, perbenzoate, decanoyl peroxide, lauryl peroxide, benzoyl peroxide and any mixtures of the said compounds. Among the aforesaid compounds, AIBN is quite particularly preferred.

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further preferred embodiment of the present invention, initiation of the polymerization is effected 15 with the use of known photoinitiators by irradiation with UV radiation or the like. Here the common, commercially available compounds such as for example benzophenone,  $\alpha$ ,  $\alpha$ -diethoxy-acetophenone, diethylaminobenzophenone, 2,2-dimethoxy-2-20 phenylacetophenone, 4-isopropylphenyl 2-hydroxy-2propyl ketone, 1-hydroxycyclohexyl phenyl ketone, isoamyl p-dimethylaminobenzoate, methyl 4 dimethylaminobenzoate, methyl o-benzoylbenzoate, benzoin, benzoin ethyl ether, benzoin isopropyl ether, 25 ether, benzoin isobutyl 2-hydroxy-2-methyl-1phenylpropan-1-one, 2-isopropylthioxanthone, dibenzosuberone, 2,4,6trimethylbenzoyldiphenylphosphine oxide, 30 bisacylphosphine oxide and others are used, and the photoinitiators can be used alone combination of two or several or in combination with one of the above polymerization initiators.

35 The quantity of the radical-forming agents can vary over wide ranges. Preferably for example quantities in the range from 0.1 to 5.0 wt.%, based on the weight of the whole composition, are used. Particularly

preferably, quantities in the range from 0.1 to 2.0 wt.%, in particular quantities in the range from 0.1 to 0.5 wt.%, each based on the weight of the whole composition, are used.

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The polymerization temperature to be selected for the polymerization is obvious to the skilled person. It is first and foremost determined by the initiator selected and the nature and manner of initiation (thermal, by irradiation, etc.). It is known that the polymerization temperature can affect the product properties of polymer. Hence in the context of the present invention polymerization temperatures in the range from 20.0°C to 100.0°C, advantageously in the range from 20.0°C to 80.0°C, in particular in the range from 20.0°C to 60.0°C are preferred. In a particularly preferred embodiment the present invention, the reaction temperature during the reaction is increased, preferably stepwise. Further, tempering at elevated temperature, for example at 100°C to 150°C towards the end of the reaction has been found advantageous.

The reaction can take place both under decreased pressure and under increased pressure. Preferably however, it is carried out under normal pressure. The reaction can take place in air or in a protective gas atmosphere, when as small as possible an oxygen content is preferably present, since this inhibits a possible polymerization.

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In a particularly preferred embodiment of the present invention, the production of the highly transparent plastic according to the invention is effected in such a manner that a homogeneous mixture of the components monomer mixture, initiator and other additives, such as for example lubricants, is produced and this is immediately filled between glass plates, whose shape is predetermined by the later use, e.g. as lenses,

spectacle lenses, prisms or other optical components. The solventless polymerization is initiated by energy input, for example by energy-rich irradiation, in particular with UV light or by heating, advantageously in the water-bath and over several hours. In this way, the optical material is obtained in its desired shape as a clear, transparent, colorless, hard plastic.

In the context of the present invention, lubricants mean additives for filled plastic materials, such as 10 compression molding materials and injection molding materials, in order to make the fillers slide more easily and hence make the compression molding materials more easily deformable. For this, for example metal soaps and siloxane combinations are suitable. Owing to 15 its insolubility in plastics, part of the lubricant migrates to the surface during processing and acts as a parting agent. Particularly suitable lubricants, nonionic fluoro surfactants, nonionic ammonium salts and 20 surfactants, quaternary phosphate esters, are described in EP 271839 A, to the disclosures wherein reference is explicitly made in the context of the present invention.

According to the invention, a highly transparent plastic with very good optical and mechanical properties is made available. Thus it has preferably a transparency according to DIN 5036 greater than 88.0%, advantageously greater than 89.0%.

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The refractive index  $n_D$  of the plastic according to the invention is preferably greater than or equal to 1.59. In general, the refractive index of a medium is dependent on the wavelength of the incident radiation and on the temperature. The information according to the invention on the refractive index therefore relates to the standard information specified in DIN 53491 (standard wavelength of the (yellow) D-line of sodium

(ca. 589 nm)).

According to the invention, the plastic preferably has an Abbé number > 36.0 according to DIN 53491. The skilled person can obtain information on the Abbé number from the literature, for example the Lexicon of Physics (Walter Greulich (publ.); Lexikon der Physik, Heidelberg; Spektrum, Akademischer Verlag, Vol. 1; 1998).

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According to a particularly preferred embodiment of the present invention, the plastic has an Abbé number > 36.0, advantageously > 37.0, in particular > 38.0.

The mechanical properties are tested by the FDA falling ball test (ANSI Z 80.1). The test is passed when the test specimen withstands the impact of a ball of 16 mm diameter undamaged: the greater the diameter of the ball with which the sample is impacted and remains undamaged, the better the material properties.

Furthermore, the plastic according to the invention is advantageously characterized by a high glass transition that at temperatures above temperature, so temperature it also retains its outstanding mechanical properties, in particular its impact resistance and its The glass transition temperature hardness. according to the invention is preferably plastic greater than 80°C, advantageously greater than 90°C, in particular greater than 95°C.

Possible fields of use for the highly transparent plastic according to the invention are obvious to the skilled person. It is particularly suitable for all uses which are indicated for transparent plastics. On account of its characteristic properties, it is above all suitable for optical lenses, in particular for ophthalmic lenses.

A further object of the present invention is a mixture containing (a) a mixture according to claim 1 and (b) one photochromic dye. For this, all 5 photochromic dyes known to the skilled person and mixtures thereof can be used. Preferably, photochromic dyes such as for example spiro(indoline)naphthoxazines, spiro(indoline)benzoxazines, spiropyrans, acetanilides, aldehyde hydrazones, thioindigo, stilbene derivatives, 10 rhodamine derivatives and anthra-quinone derivatives, benzofuroxans, benzopyrans, naphtho-pyrans, organometallic dithiozonates, fulgides and fulgimides are used.

- 15 From these mixtures, photochromic materials can be produced which are used for example as lenses, preferably optical lenses, glass panes or glass inserts.
- The following examples and the comparison example serve for illustration of the invention, without it being intended that any restriction should result from this.

## Examples

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Synthesis of the Thiomethacrylate Mixture 75.36 g of 1,2-ethanedithiol are weighed into a conical flask with a blanket gas inlet and stirred, and 416.43 g of 13% NaOH solution are fed in within 30 minutes at 25-30°C with water cooling. A brownish, clear solution is formed.

178.64 g of methacrylic anhydride and the Na thiolate 10 solution are now fed in parallel within 45 minutes at input temperature into the previously the desired prepared and stirred acetic acid/water in the reaction flask. In general, the flask contents cool by ca. 2°C at the start of the addition, and after ca. 5-10 minutes a 15 slightly exothermic reaction begins, i.e. the mixture is now cooled appropriately, in order to maintain the desired reaction temperature (35°C). After the end of the addition, the mixture is stirred for a further 5 minutes at 35°C and then cooled to ca. 25°C with 20 stirring.

The mixture is transferred to a separating funnel, separated and the lower, aqueous phase drained off. For the workup, the organic phase is transferred to a conical flask and stirred for ca. 15 minutes with  $^{\circ}$ Dowex M-31, and then the ion exchanger is filtered off.

30 The somewhat cloudy to almost clear crude ester solution is now stabilized with 100 ppm of HQME and concentrated on the rotary evaporator at max. 50°C. The colorless final product is filtered at room temperature (20-25°C). Ca. 140 g of colorless, clear ester are obtained.

Production of Prepolymer: Reaction of 6.84 g of the thiodi-(meth)acrylate and 0.36 g of DMDO in the presence of an amine as catalyst analogously to EP 284374

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For the production of a polymer based on an oligomeric thiodimethacrylate, example 7.2 for q of prepolymer, 2.4 g of styrene, 2.4 g of 10-times ethoxylated bisphenol A di(meth)acrylate, 0.1 hydroxyethyl methacrylate, 36 mg of a UV initiator such as for example Irgacur 819 and 24 mg of tert.-butyl peroctoate or similar initiators (see Example 1) are mixed. The homogeneous casting resin mixture is fed into a suitable mould and cured within 10 minutes in a UV curing unit with 1200 W mercury high pressure lamps. It is then tempered for a further 2 hrs at ca. 120°C in the oven.

Expt.	System	Refract./Abbé No. DIN 53491 589 nm		Odor	Trans- missio n	FDA falling ball test diameter ANSI	Mean ball diameter in mm test passed
	<u> </u>		ļ			Z80.1	
Examples							
B 1	PLEX 6931/DMDO prepolymer co styrene co E10BADMA = 60:20:20 plus 1% HEMA	1.5939	38.2	no	89	passed	18
Comparison Examples							
VB I	PLEX 6931 co DMDO co styrene co E10BADMA = 57:3:20:20 plus 1% HEMA (#)	-	-	yes	-	-	-
VB II	PLEX 6931 co styrene co E10BADMA = 60:20:20 plus 1% HEMA	1.5959	34.9	no	89	passed	16
VB III	PLEX 6931/DMDO prepolymer co styrene = 70:30 plus 1% HEMA	1.6089	29.6	no	89	passed	16

20 Plex 6931 O: reaction product from methacrylic anhydride and ethanedithiol from DE 316671 E10BADMA: ethoxylated bisphenol A dimethacrylate with

ethoxylation level of ca. 10 DMDO: dimercaptodioxaoctane

25 HEMA: hydroxyethyl methacrylate

(#): because of the odor problem, no further analytical tests were made.

The mixture according to the invention (B 1) is odorless. The comparison example VB I did not pass this

test, hence it was not further investigated.

However, at comparable refractive index (of B1 with VB II and VB III) the Abbé number was better with the mixture according to the invention. In addition, the mixture according to the invention came out considerably better in the falling ball test.